

Modified Bitumen Emulsion for Coating Fertilizers

El-Hady, O.A. and Camilia, Y. El-Dewiny

Soils and Water Use Department, National Research Center, Dokki, Cairo, Egypt.

Abstract: One of the main objectives of our research program is to produce slow and at the same time controlled release fertilizers using bitumen as a local and a cheap coating material. A complex fertilizer 20:20:20 was coated with a modified bitumen emulsion (50% active material) prepared using bitumen of penetration 180/200 (a product of Alexandria Petroleum Company); the cationic emulsifier tri-ethanol ammonium hydroxide; Quasar El-Sagha (Fayoum) bentonitic clay and the emulsified polymer (50% active material) polyvinyl acetate butyl acrylate. Four coating fertilizer ratios were examined namely; 0:100; 5:95; 10:90 and 15:85. Dissolution rate and released N,P and K after soaking the fertilizer in deionized water for different periods up to 120 days under constant temperature (100 °F i.e. 37.8C°) indicate the possibility of using bitumens as basic materials for producing effective coatings for water soluble fertilizers . Coating thickness influences the release rate and longevity of the product. Costs of preparation are economically justified.

Key words: Controlled release, fertilizer coating, bitumen emulsion, dissolution rate, Released nutrients

INTRODUCTION

Application of regular soluble fertilizers -particularly to sandy soils that have very little ability to maintain soluble nutrient elements- is difficult to practice. Loss of fertilizers by leaching and deep percolation, frequent fertilizer applications; low crop use efficiency; pollution of ground water and eutrophication of natural surface waters besides increasing fertilization costs and labour expenses are expected. Therefore, specific fertility management practices will be required to grow crops on these sandy soils. Application of controlled release fertilizers (CRF) to release nutrients in small quantities over an extended period of time may solve most of the aforementioned problems. Numerous researches have demonstrated the superiority of slow release fertilizers in experiments with forage, grains, vegetables and fruit trees grown in field and green house. They reported that single application of slow release fertilizer gave maximum yield production and seasonal distribution of growth similar to those obtained with the multiple (3-5) application of urea or ammonium nitrate^[19,10,13,14,18,11,3,16].

In recent years, due to public demands for environmental protection and the need for increasing production efficiency, great efforts in developing and manufacturing slow and at the same time CRF became necessary. Resin coating technology has been adapted to produce CRF. With coating, the inner soluble nutrient elements will be released slowly to be taken by the growing plant as it is required^[19]

Producing CRF using bitumens as local and cheap coating materials is of the main objectives of our research program. Previous studies by the others^[6,7,5] encouraged us to continue the work. That is the aim of the present study.

MATERIALS AND METHODS

MATERIALS:

A- Fertilizer: A complex water-soluble fertilizer 20:20:20; that ordinary used in fertilization of new reclaimed sandy areas. It contains 20% of each of N, P₂ O₅ and K₂ O.

B- Coating material: The examined coating material is a modified bitumen emulsion (50% active material). Used materials for preparation were bitumen of penetration 180/200" a product of Alexandria Petroleum Company"; the cationic emulsifier tri ethanol ammonium hydroxide; Quasar El-Sagha (Fayoum) bentonitic clay and the emulsified polymer polyvinyl acetate butyl acrylate (50% active material) that locally prepared at NRC^[8].

METHODS:

1- Preparation of the coating:

a- Bitumen emulsion 50% active material was firstly prepared as follows: Water was warmed up to 70 °C. The bentonitic clay (5% by weight) and the emulsifier (2% by weight) were added while stirring, followed by hot bitumen (135 °C). Stirring was continued until the emulsion became completely homogenous.

b- The emulsified polymer PVAcBuA emulsion (50% active material) was added to the emulsified bitumen just before using in the coating process, taking into consideration that bitumen polymer ratio was 9:1. Stirring was continued until the emulsion became completely homogenous^[17]

2-Coating process:

The prepared coating material was sprayed on the fertilizer using a modified horizontal rotary drum with pan granulator. It provides a homogenous (with respect to particle size), dense, mass of sized particles in random motion so that highly uniform coating can be applied to each particle. General, description of the apparatus was previously mentioned^[4] Four different coating fertilizer ratios were examined i.e. 0:100, 5:95, 10:90 and 15:85.

Evaluation of the prepared fertilizers:

1- Dissolution rate: Uncoated and coated fertilizers were evaluated for their releasing rate as follows: A weight that contains five grams of the fertilizer was enclosed in a nylon screen bag. The bag was inserted vertically in a glass jar containing 25 ml deionized water. Jars were covered and placed in an incubator at 100 °F (37.8 °C) for the appropriate time interval. Dissolved nutrients were determined by calculating the weight loss from each sample as a function of time. Data of the mean value of four replications were graphically plotted versus time.

2- Released nutrients as a function of time: Samples from the remained solutions were taken after of 1, 2, 3, 5, 7, 14, 21, 28, 35, 42, 49, 56, 60, 90 and 120 days. Determination of N, P and K were carried out using methods described by Bremner and Mulvany^[1] for nitrogen and cottenie *et al*^[2] for the other two nutrients.

RESULTS AND DISCUSSIONS

Dissolved amounts of the uncoated and coated fertilizers after immersing in deionized water for different periods up to four months (120 days) under constant temperature (100 °F) are illustrated in figure 1. It is worthy to note that each value is the mean of four replications. Released N, P and K are shown in figures 2, 3 and 4, respectively. A considerable amount of the uncoated fertilizer was released in the first day to reach more than 75% of the fertilizer and its nutrients. The disappearance of undissolved fertilizer was observed at the third day. On the other hand, coated granules released their nutrients much more slowly. Considerable amounts of dissolved fertilizer and nutrients are still released. The higher the coating fertilizer ratio i.e. the thicker the coating is, the

Table 1: Time (days) required for the fertilizer to release 25%, 50% and 75% of its inner content under standard conditions (100 F° in water).

Coating : Fertilizer	% of release		
	25	50	75
Dissolved fertilizer			
0:100	<1	<1	1
5:95	3	25	54
10:90	10	37	85
15:85	16	50	>120
Released N			
0:100	<1	<1	1
5:95	3	24	54
10:90	9	36	81
15:85	15	50	>120
Released P			
0:100	<1	<1	<1
5:95	4	23	50
10:90	10	34	73
15:85	16	50	>120
Released K			
0:100	<1	<1	<1
5:95	3	25	56
10:90	10	36	80
15:85	15	48	>120

slower is the release of the fertilizer nutrients and the longer is the period of release, referring to the possibility of long duration of the fertilizer in the soil. The statistical relations between the percentages of dissolved fertilizer or released nutrients (y) and time in days (x(as affected by coating thickness are inserted in the figures. Regression equations and correlation coefficients denote highly significant relations. Taking these relations into consideration, the time in days required for the fertilizer to release different percentages of its inner content i.e. 25, 50 and 75% under the conditions of the study are shown in Table 1. Data indicate the aforementioned fact that coating thickness influences the release rate and longevity of the product.

In general, there are two release patterns for the coated fertilizer pellet to release soluble nutrient elements. An immediately and relatively rapid release through a few

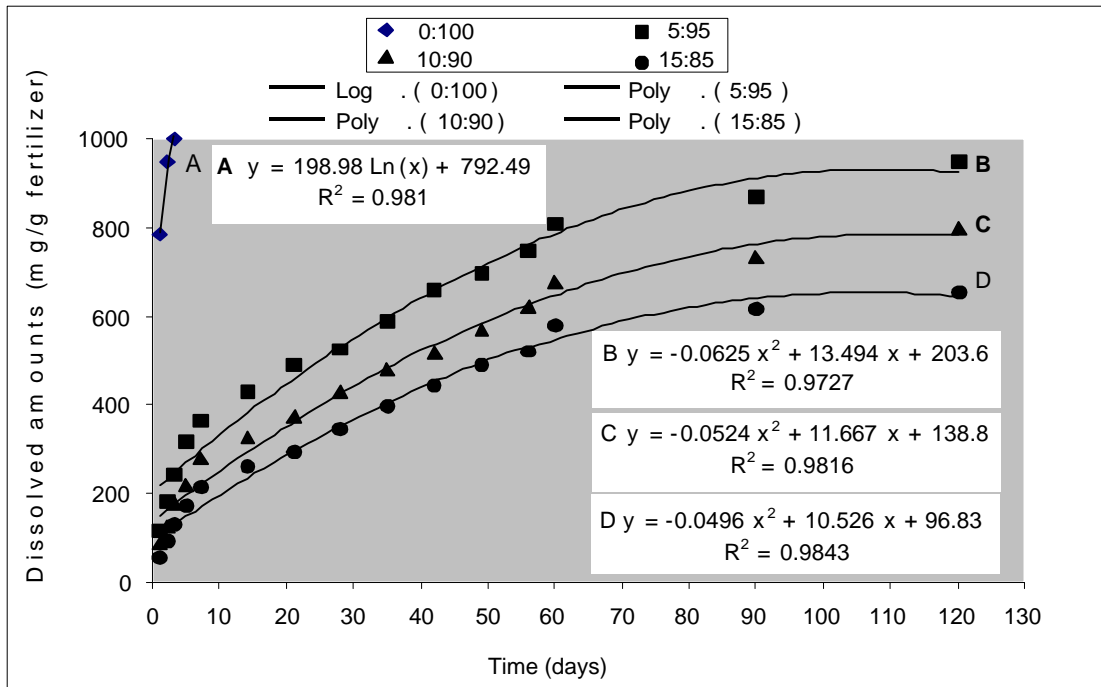


Fig. 1: Dissolved amounts (mg/g fertilizer).

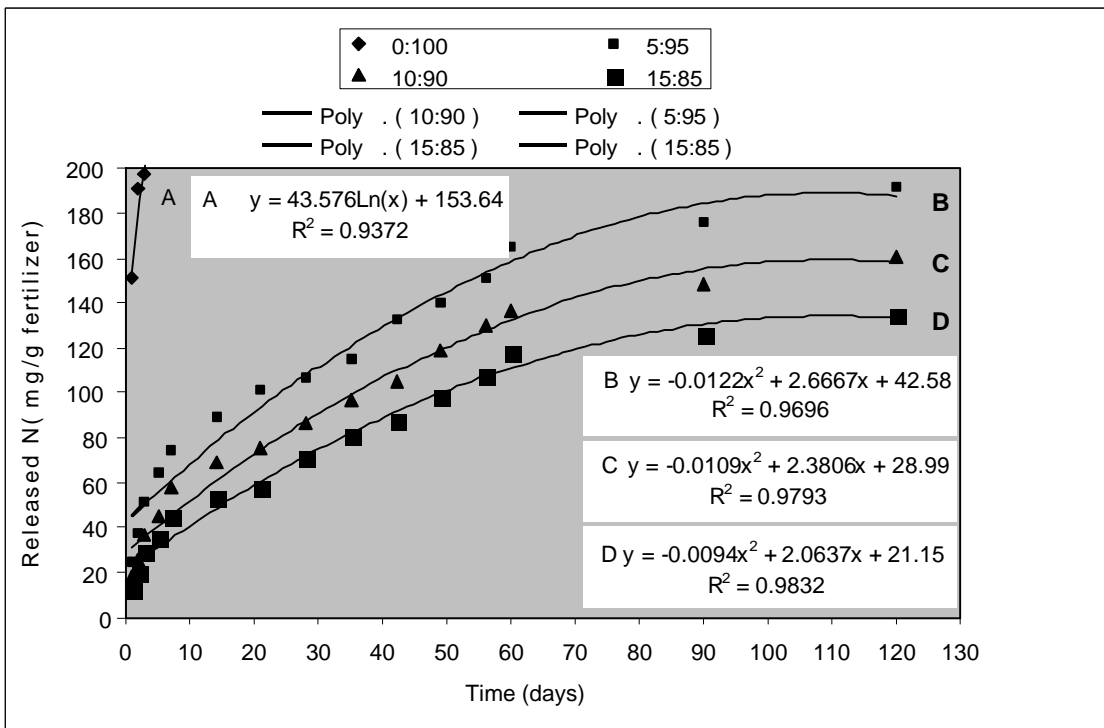


Fig. 2: Released N (mg/g fertilizer).

holes or many microscopic pores originally present in the coating and little release for an extended period of time^[11,19]. Therefore, controlling the thickness as well as

using multiple coating techniques can control the rate of nutrient element release. Recent development in coating technology allows duration of nutrient elements from

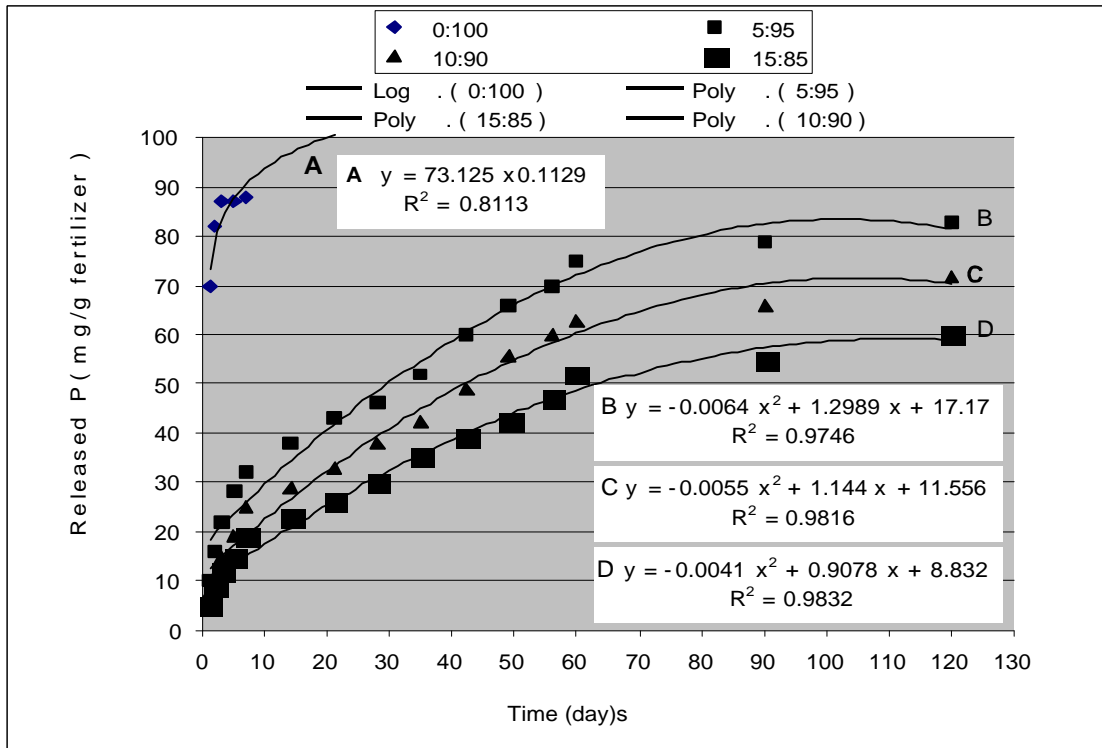


Fig. 3: Released P (mg/g fertilizer).

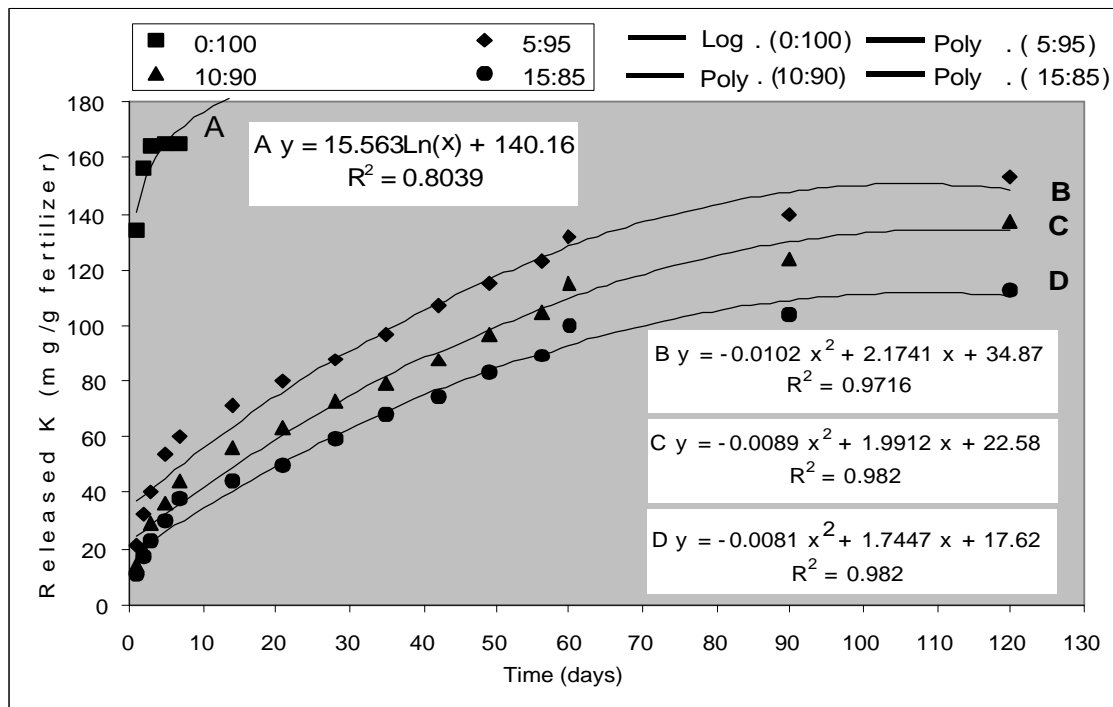


Fig. 4: Released K (mg/g fertilizer).

20 to 700 days (measured in water at 25 °C). Nutrient element released from coated fertilizers could match crop

uptake during the whole growing period, if a well designed mixture of coated fertilizers is used. This would

allow coated fertilizers to have a higher crop use efficiency than that obtained from regular fertilizers. Researches comparing regular and coated fertilizers have demonstrated how coatings improve crop uptake efficiency^[9] reported that poly-olefin or lignin-coated N fertilizers were particularly effective in minimizing N losses and enhancing N use efficiency in sandy soils. In the case of P, reduced contact between the fertilizer P and the soil extended the time that P was available to plants^[12] observed that P recovery rates were three to five times higher with polyolefin-coated double super phosphate and multi-super phosphate than with uncoated ones. In calcareous soils with high P-fixation capacity, where application of uncoated P fertilizers did not significantly increase soil available P, coated P fertilizer did. Coated trace nutritive elements such as Zn appeared superior to the regular ones. Polyolefin coated Zn increased Zn uptake in a Zn deficient soil resulting in a significant yield increase compared with that obtained using a soluble Zn form^[15]

Costs: It is interesting to note that the main components of the examined coating are locally prepared. Producing one ton of this coating costs \approx 1000 L.E. (prices of 2005). With this respect, one ton of the regular soluble fertilizer (2750 L.E.) needs only \approx 53,111 and 176 L.E. for the coating material to produce CRF_s having the three coating fertilizer ratios, 5:95, 10:90 and 15:85, respectively. This will be economically justified if compared with other types of coatings.

Obtained results push us to continue the work with other coating formulations based on asphalt and soluble fertilizers having both macro and micro-nutrients at the same time.

REFERENCES

1. Bremner, J.M. and C.S. Mulvany, 1982. Nitrogen-total. In A.L. Page, *et al.*, (ed.) *Methods of Soil Analysis*. Part 2, 2nd ed. Agron. Monogr, ASA and SSSA, Madison, W.I. pp 595-624.
2. Cottenie, A., M. Verloo, G. Velghe and R. Camerlynck, 1982. *Chemical analysis of plants and soils*. Laboratory of Analytical and Agro chemistry, State Univ. Ghent, Belgium.
3. Diez, J.A., R. Caballero, A. Bustos, R. Roman, M.C. Cartagena and A. Vallejo, 1996. Control of nitrate pollution by application of controlled released fertilizer (CRF), Compost and an optimized irrigation system. *Fertilizer Research* 43:191-195.
4. El-Aila, H.I. and M. Abou-Seeda, 1996. Studies on slow release fertilizer: I. The utilization of controlled release and traditional nitrogen fertilizers by wheat. *J.Agric. Sci. Mansoura Univ.*, 21: 4161-4178.
5. El-Hady, O.A., A.A. Abd El-Kader and N.F. Ghaly, 2003_a. Fertilizer bitumen combination for improving the fertility status of desert soils. *Egypt, J. Appl. Sci.*, 18: (4B) 737-750.
6. El-Hady, O.A. and N.F. Ghaly, 2000. Controlled release fertilizers based on asphalt. *Egyptian Soil Sci. Soc. (ESSS). Golden jubilee Congress, Oct.23-25, 2000, Cario. paper no.p.3.5.*
7. El-Hady, O.A. and N.F. Ghaly, 2004. Effective bitumen coating for water soluble fertilizers. *The 7th Int. Conf. On "Petroleum and the environment."* EPRI, V.E.A. and I.S.A. 27-29 March, 2004, Cairo, Egypt.
8. El-Hady, O.A., N.F. Ghaly and A.M. Zaghoul, 2003_b. Polymerized bitumen emulsions in remediation of heavy metal contaminated soils. *Egypt, J. Soil Sci* 43(4):529-545.
9. Garcia, M.C., J.A. Diez, A. Vallejo, L. Garcia and M.C. Cartagena, 1997. Effect of applying soluble and coated phosphate fertilizers on phosphate availability in calcareous soils and on P absorption by a ryegrass crop. *J. Agric. Food Chem.* 45:1931-1936.
10. Jian Rong, F., Z. Yuan Hong and J. Lina, 2001. Use of controlled release fertilizer for increasing N efficiency of direct seeding rice. *Pedosphere*, 11:333-339.
11. Ko, B.S. Cho, Y.S. and H.K. Rhee, 1996. Controlled release of urea from resin - coated fertilizer particles. *Ind. Eng. Chem. Res.* 35: 250-257.
12. Nazyo, M., H. Kurosaki and S. Yamasavi, 1998. Characteristics of P supply poly - olefin - coated P fertilizers in soil. *Soil Sci. Plant Nutrition* 43:1073-1078.
13. Paramasivam, S. and A.K. Alva, 1997_a. Leaching of nitrogen from controlled release nitrogen fertilizers. *Commun. Soil Sci. Plant Anal.* 28: (178518)1663-1674.
14. Paramasivam, S. and A.K. Alva, 1997_b. Nitrogen recovery from controlled release fertilizers under intermittent leaching and cycles. *Soil Sci.*, 162 (6) : 447-453.
15. Rico, M.I., J.M. Alvarez and J.I. Mingot, 1996. Efficiency of zinc ethylene di-amine tetra acetate and zinc lingo-sulfonate soluble and coated fertilizers for maize in calcareous soils. *J. Agric. Food Chem.* 44: 3219-3223.

16. Sharon, G., K. Kochbo and Yoram, 1990. Studies on slow release fertilizers: II.A method for evaluation of nutrient release rate from slow releasing fertilizers. *Soil Sci.*, 150: 446-450.
17. Sitz, G.E., Maysville, Roger, K. Chatterjee and Columbus, 1991. Process for the production of polymer modified asphalts and asphalt emulsions. M.S.Patent,P no. 5,109 , 610 .
18. Wang, F.L. and A.K. Alva, 1996. Leaching of nitrogen from readily soluble and slow release fertilizers in sandy soils. *Soil Sci. Soc. Am. J.* 60: 1454-1458.
19. Wen, G., M. Tomoko, T. Yamamoto, J. Chikushi and M. Inoue, 2001. Nitrogen recovery of coated fertilizers and influence on peanut seed quality for peanut plants grown in sandy soil. *Commun. Soil Sci. Plant Anal.* 32: 3121-3140.